

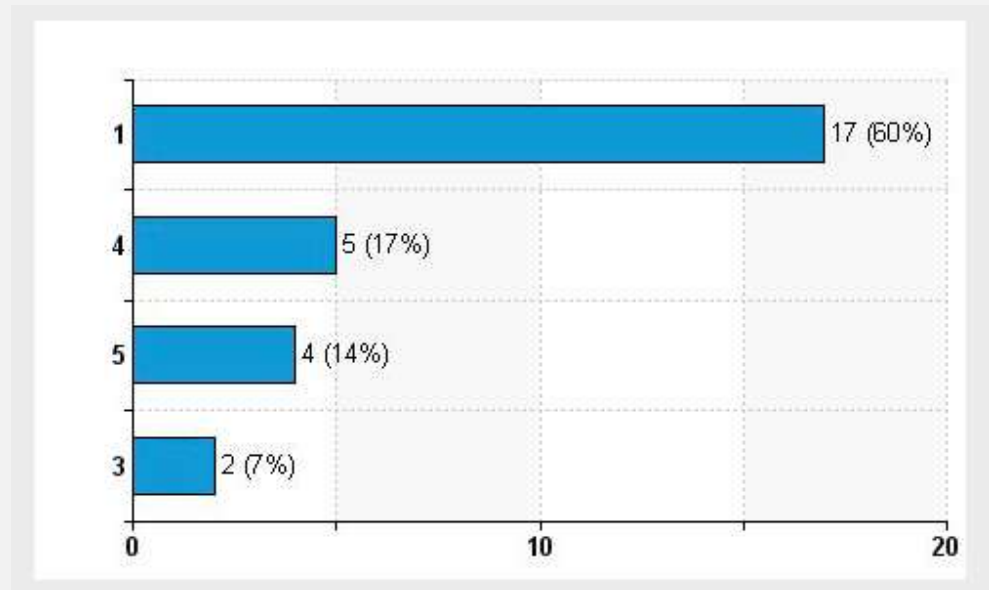


Measurement

Dr. Venkat Kaushik
PHYS 211, Fall 2015

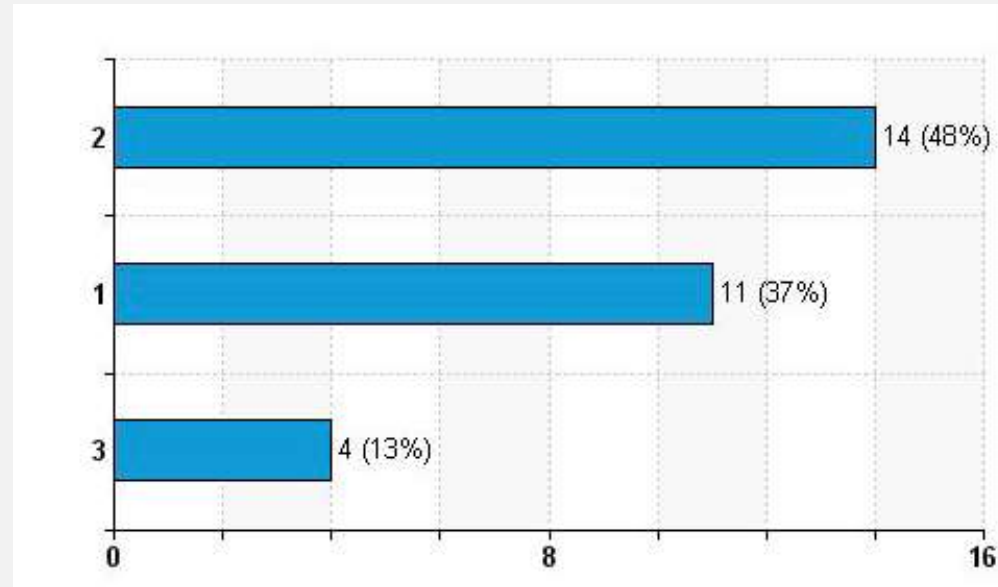
Pulse Question 1: (Anonymous)

- What best describes your view of Physics?
 1. Enthusiastic
 2. Don't care, Nonchalant
 3. Terrified, just allow me to get a C professor!
 4. Confused, do not know where to make of it.
 5. None of the above



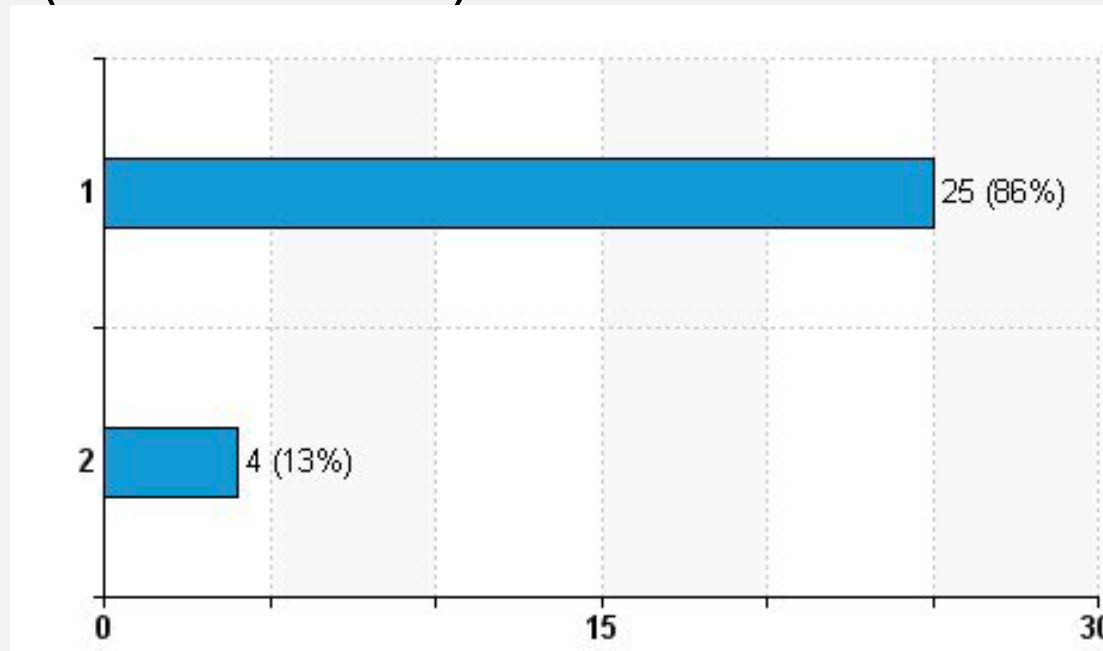
Pulse Question 2: (Anonymous)

- What best describes your current GPA?
 1. A, A+
 2. B, B+
 3. C, C+
 4. D, D+



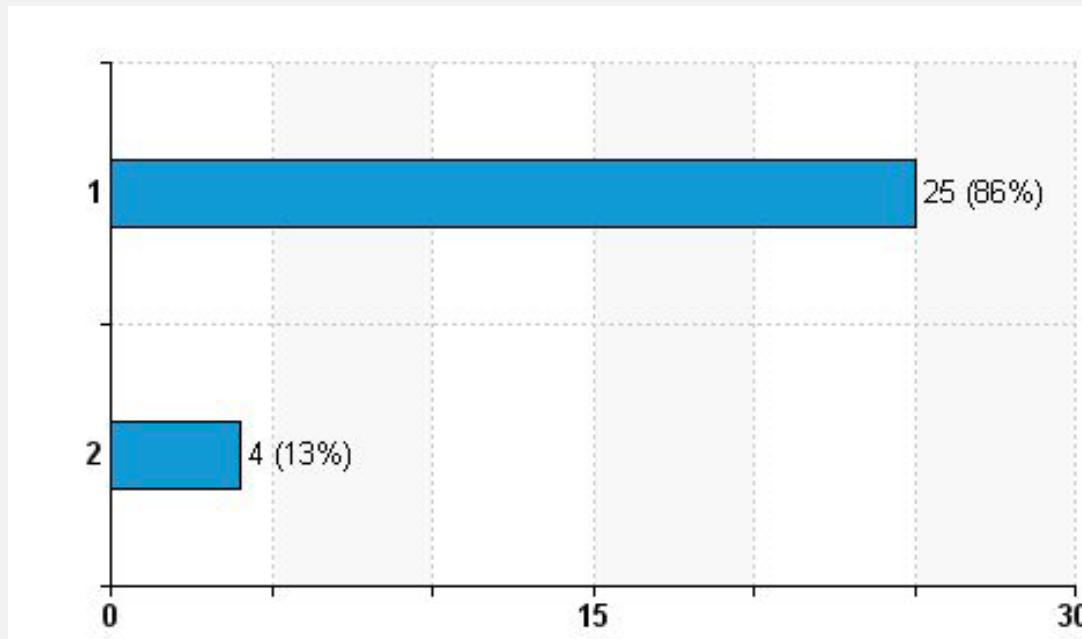
Pulse Question 3: (Anonymous)

- What grade do you wish to make in PHYS 211?
 1. A
 2. B
 3. C
 4. D, Other (Do NOT Care)

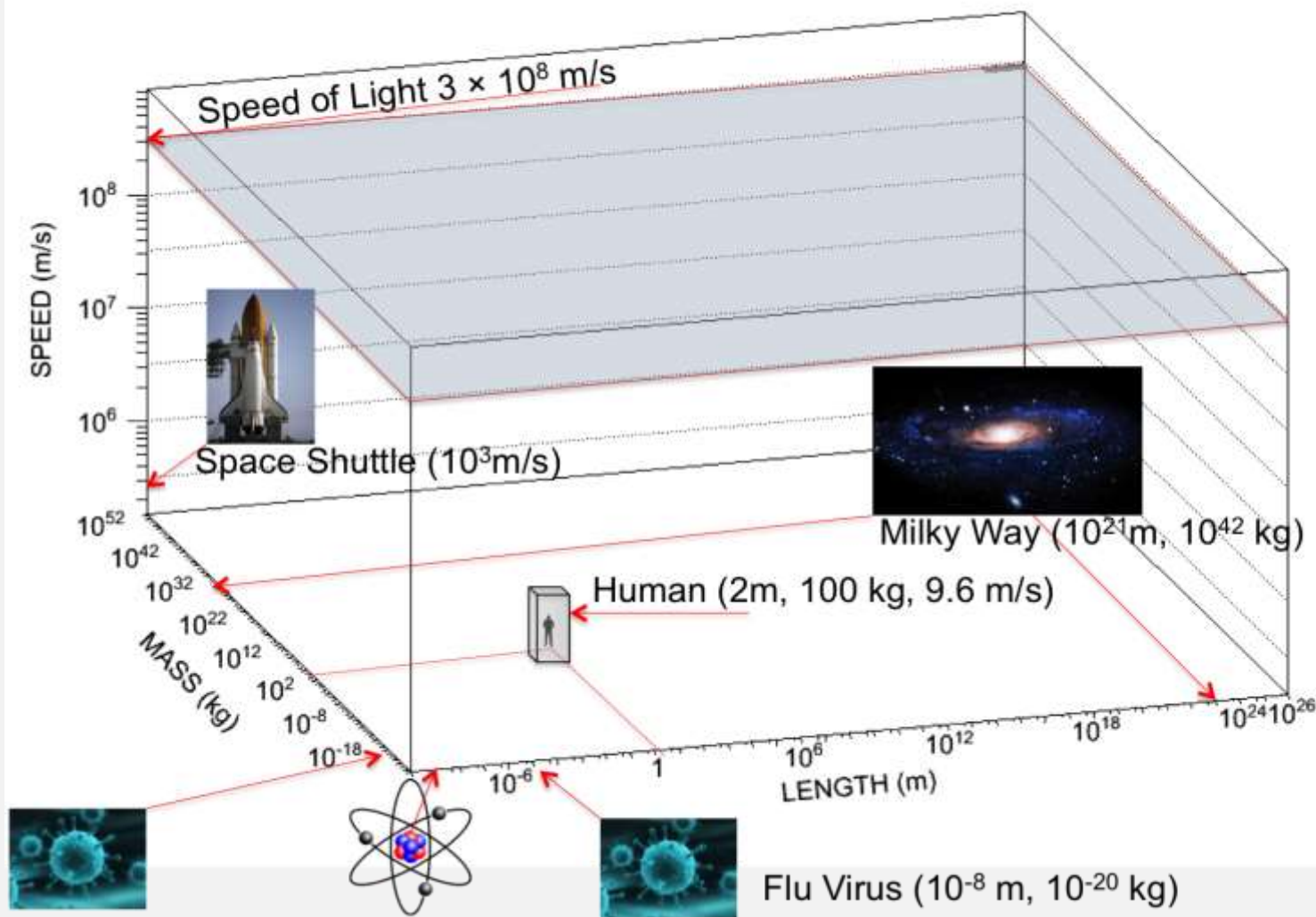


Clicker Question 1

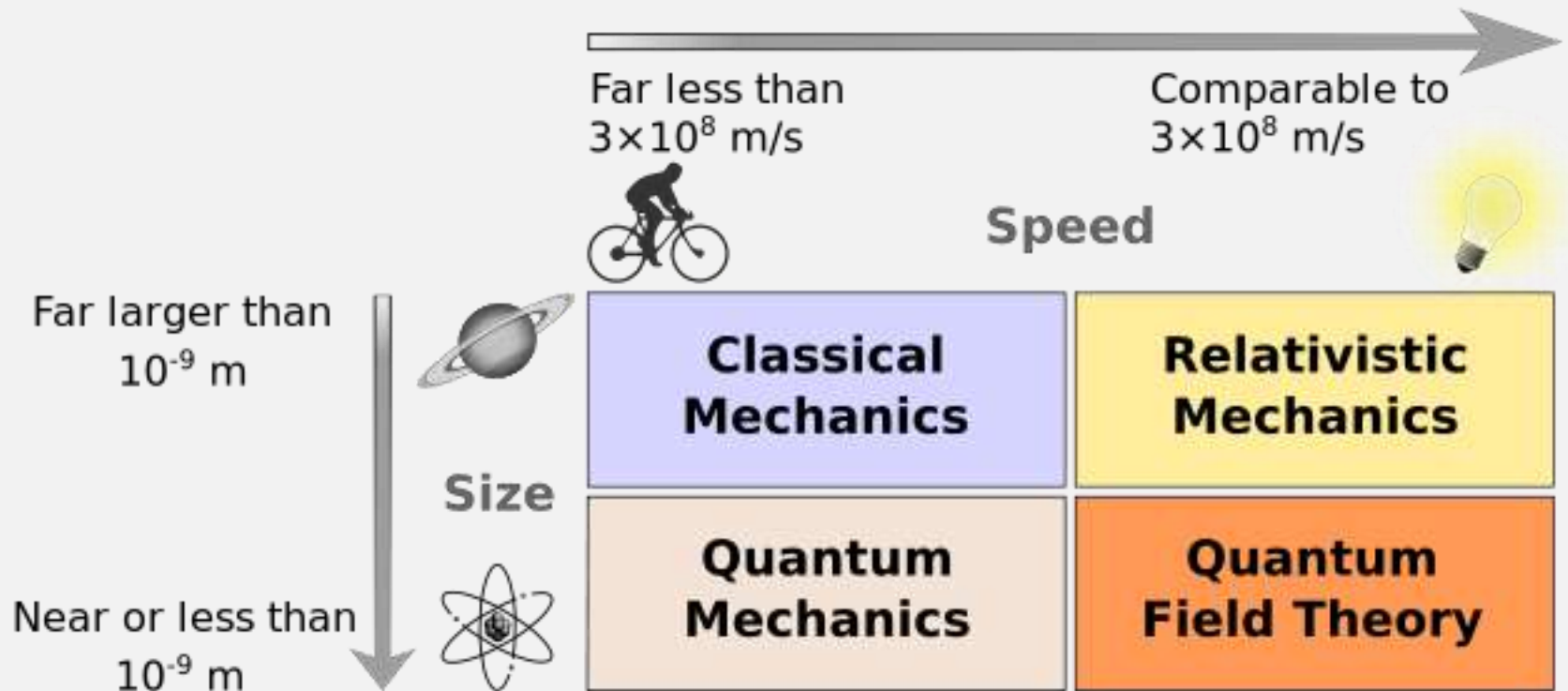
- I have read the syllabus (one or more times)
 1. YES, I HAVE
 2. NO, I HAVE NOT



Our Limited Access To Reality

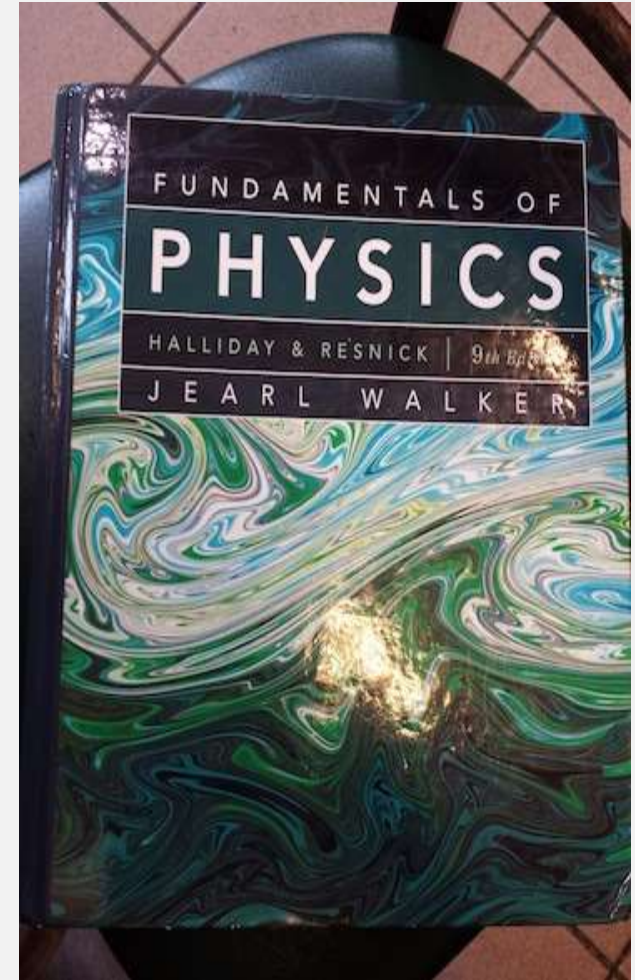


Classical Mechanics



Physical Quantities

- Physical Quantity is defined as
 - a **quantifiable property** assigned to a phenomenon, body or a substance
- Take your text-book as an example
 - It's an object that has certain **dimensions**
 - You can hold it and feel it's **heavy**
- In Phys 211, 212 you will encounter some
 - Fundamental physical quantities: Mass, Length, Time, Temperature, Electric charge
 - Derived physical quantities: Speed, Acceleration, Electric/Magnetic Fields



Units

- **Mass**
 - 1 kilogram (kg), or pound (lb)
- **Length**
 - 1 meter (m) or mile (mi)
- **Time**
 - 1 second (s)
- **Temperature**
 - 1 degree Celsius ($^{\circ}\text{C}$) or Fahrenheit ($^{\circ}\text{F}$)
- **Electric charge**
 - 1 Coulomb (C)
- **Speed**
 - 1 meter/second (m/s)
 - 1 mile/hour (mi/h)
- **Acceleration**
 - 1 meter per second squared (m/s^2)
- **Force**
 - $1 \text{ kg m/s}^2 = 1 \text{ N}$ (newton)
- **Energy**
 - $1 \text{ kg m}^2/\text{s}^2 = 1 \text{ J}$ (joule)
- **Electric Field**
 - 1 N/C (newton per coulomb)

Unit: A unique name assigned to a measure of that physical quantity

International System (SI)

- Also called as “metric system” (1971)
 - Adopted seven base units at the conference of weights and measures
 - The kilogram (kg) – unit of measurement of mass
 - The meter (m) – unit of measurement of length
 - The second (s) – unit of measurement of time
 - The ampere (A) – unit of measurement of electric current
 - The kelvin (K) – unit of measurement of thermodynamic temperature
 - The mole (mol) – unit of measurement of amount of substance
 - The candela (cd) – unit of measurement of luminous intensity
- How to express very large or very small numbers?
 - Scientific notation: $5670 \text{ N} = 5.67 \times 10^3 \text{ N}$ (or 5.67E3 on calculator)
 - Use of prefixes: $5670 \text{ N} = 5.67 \text{ kN}$ (kilo-newton) 1 kilo = 10^3

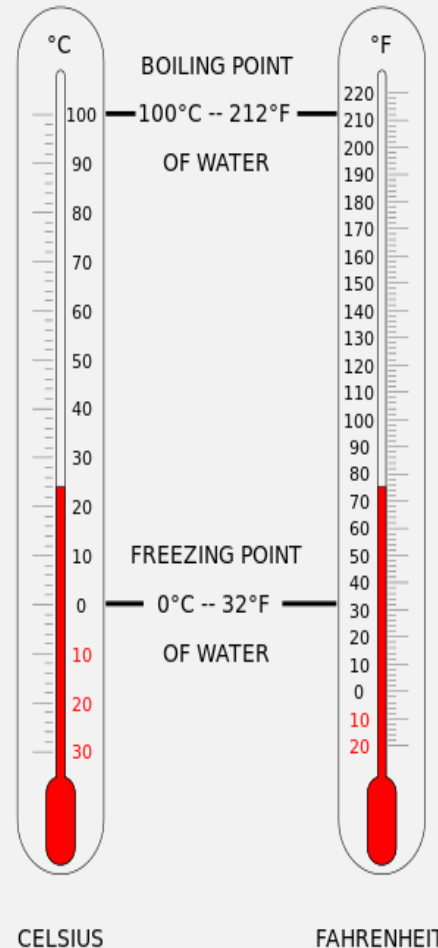
Converting Units

- Converting degree Celsius ($^{\circ}\text{C}$) to degree Fahrenheit ($^{\circ}\text{F}$)

Let C denote the value (in Celsius) corresponding to the value F (in Fahrenheit) of the same temperature. Notice their linear relationship. Assume a slope (m) and an intercept (b). Looking at the scales $(C,F) = \{ (0, 32), (100, 212) \}$ must lie on the line given by the equation $C = mF + b$. Solving for m and b , we get

$$m = 5/9, \quad b = -(5/9) \times 32$$

$$C = \frac{5}{9}(F - 32)$$



Reading Assignment

- Length
 - <http://www.nist.gov/pml/wmd/metric/length.cfm>
- Mass
 - <http://www.nist.gov/pml/wmd/metric/mass.cfm>
- Time
 - <http://www.nist.gov/pml/wmd/metric/time.cfm>
- SI Units
 - <http://www.nist.gov/pml/wmd/metric/si-units.cfm>